

Did you know?

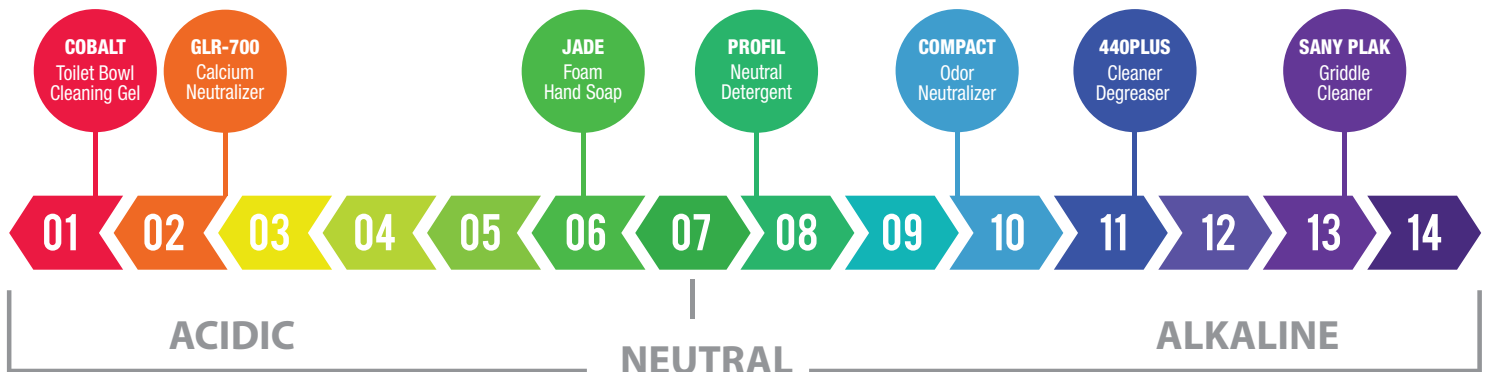
Why are we talking about pH in cleaning products?

The pH, which stands for hydrogen potential, is a measure of the acidity or alkalinity of an aqueous solution.

A pH scale is used to determine the degree of acidity and alkalinity of a cleaning product.

The pH scale is divided into steps from 0 to 14, where 0 represents the highest degree of acidity and 14 the lowest degree of acidity. The central point of the scale (7) is neutral. The pH of distilled water is 7, which is considered the purest possible neutral solution.

Product pH scale



Understanding this scale makes the cleaning job easier and avoids using the wrong product for the job at hand.

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For example:

For the removal of lime, calcium, rust, juice or coffee:

An acid cleaner with a pH between 1 and 7 should be used. Attention these require special care to avoid damaging the surfaces.

For painted surfaces, waxed floors and wood floors:

A neutral cleaner with a pH between 6 and 8 is recommended.

For hand cleaning:

A product with a neutral pH between 7 and 8.

For windows, mirrors, cars:

To avoid discoloration and unwanted streaks an alkaline detergent with a pH between 8 and 11 is recommended.

In general, vegetal, animal or human stains can be removed with an alkaline product and for the removal of accumulation of mineral, metallic or saline substances with an acidic product.

Be careful, it is important not to mix the products. In case of doubt, consult one of our in-store specialists.